

DAFTAR PUSTAKA

- Apriliani, A. (2010). Pemanfaatan Arang Ampas Tebu sebagai Adsorben Ion Logam Cd, Cr, Cu dan Pb dalam Air Limbah. *Repositoy UIN*, 1–91.
- Bernhard, A. (2010). The nitrogen cycle: processes. *Nature Education Knowledge*, 2(2), 1–8. <https://www.nature.com/scitable/knowledge/library/the-nitrogen-cycle-processes-players-and-human-15644632/>.
- Cechinel, M. A. P., Ulson De Souza, S. M. A. G., & Ulson De Souza, A. A. (2014). Study of lead (II) adsorption onto activated carbon originating from cow bone. *Journal of Cleaner Production*, 65, 342–349. <https://doi.org/10.1016/j.jclepro.2013.08.020>
- Cengeloglu, Y., Tor, A., Ersoz, M., & Arslan, G. (2006). Removal of nitrate from aqueous solution by using red mud. 51, 374–378. <https://doi.org/10.1016/j.seppur.2006.02.020>
- Chen, B., Chen, Z., & Lv, S. (2011). A novel magnetic biochar efficiently sorbs organic pollutants and phosphate. *Bioresource Technology*, 102(2), 716–723. <https://doi.org/10.1016/j.biortech.2010.08.067>
- Dąbrowski, A. (2001). Adsorption - From theory to practice. *Advances in Colloid and Interface Science*, 93(1–3), 135–224. [https://doi.org/10.1016/S0001-8686\(00\)00082-8](https://doi.org/10.1016/S0001-8686(00)00082-8)
- Divband Hafshejani, L., Hooshmand, A., Naseri, A. A., Mohammadi, A. S., Abbasi, F., & Bhatnagar, A. (2016). Removal of nitrate from aqueous solution by modified sugarcane bagasse biochar. *Ecological Engineering*, 95, 101–111. <https://doi.org/10.1016/j.ecoleng.2016.06.035>
- Fatikhah, A. (2019). *Estimasi Beban Pencemar Nitrogen Berdasarkan Pola Penggunaan Pupuk Pada Lahan Pertanian di Kecamatan Pakem, Yogyakarta*. Universitas Islam Indonesia.
- Gai, X., Wang, H., Liu, J., Zhai, L., Liu, S., Ren, T., & Liu, H. (2014). Effects of feedstock and pyrolysis temperature on biochar adsorption of ammonium and nitrate. *PLoS ONE*, 9(12), 1–19. <https://doi.org/10.1371/journal.pone.0113888>
- Gao, F., Xue, Y., Deng, P., Cheng, X., & Yang, K. (2015). Removal of aqueous ammonium by biochars derived from agricultural residuals at different pyrolysis temperatures. *Chemical Speciation and Bioavailability*, 27(2), 92–97. <https://doi.org/10.1080/09542299.2015.1087162>
- Hamuna, B., Tanjung, R. H. R., Suwito, S., & Maury, H. K. (2018). Konsentrasi Amoniak, Nitrat Dan Fosfat Di Perairan Distrik Depapre, Kabupaten Jayapura. *EnviroScientiae*, 14(1), 8. <https://doi.org/10.20527/es.v14i1.4887>
- Hu, X., Zhang, X., Ngo, H. H., Guo, W., Wen, H., Li, C., Zhang, Y., & Ma, C.

- (2020). Comparison study on the ammonium adsorption of the biochars derived from different kinds of fruit peel. *Science of the Total Environment*, 707, 135544. <https://doi.org/10.1016/j.scitotenv.2019.135544>
- Iftekhar, S., Ramasamy, D. L., Srivastava, V., Asif, M. B., & Sillanpää, M. (2018). Understanding the factors affecting the adsorption of Lanthanum using different adsorbents: A critical review. *Chemosphere*, 204, 413–430. <https://doi.org/10.1016/j.chemosphere.2018.04.053>
- Jindo, K., Mizumoto, H., Sawada, Y., & Sonoki, T. (2014). Physical and chemical characterizations of biochars derived from different agricultural residues. *Biogeosciences Discussions*, 11(8), 11727–11746. <https://doi.org/10.5194/bgd-11-11727-2014>
- Jung, K. . (2015). *Kinetic study on phosphate removal from aqueous solution by biochar derived from peanut shell as renewable adsorptive media*. <https://doi.org/10.1007/s13762-015-0766-5>
- Karimi, K., Kheradmandinia, S., & Taherzadeh, M. J. (2006). Conversion of rice straw to sugars by dilute-acid hydrolysis. *Biomass and Bioenergy*, 30(3), 247–253. <https://doi.org/10.1016/j.biombioe.2005.11.015>
- Katal, R., Sharifzadeh, M., Taher, H., & Esfandian, H. (2012). Journal of Industrial and Engineering Chemistry Kinetic , isotherm and thermodynamic study of nitrate adsorption from aqueous solution using modified rice husk. *Journal of Industrial and Engineering Chemistry*, 18(1), 295–302. <https://doi.org/10.1016/j.jiec.2011.11.035>
- Legowo, S. (2017). Pertumbuhan Kultur Tunggal Dan Campur Jamur Pelapuk Putih (*Phanerochaete chrysosporium* dan *Trametes versicolor*) Pada Proses Biodelignifikasi Jerami Padi. In *Universitas Muhammadiyah Surakarta* (Vol. 4).
- Lin, L., Lei, Z., Wang, L., Liu, X., Zhang, Y., Wan, C., Lee, D., & Tay, J. H. (2013). Adsorption mechanisms of high-levels of ammonium onto natural and NaCl-modified zeolites. *SEPARATION AND PURIFICATION TECHNOLOGY*, 103, 15–20. <https://doi.org/10.1016/j.seppur.2012.10.005>
- Ma, Z., Li, Q., Yue, Q., Gao, B., Li, W., Xu, X., & Zhong, Q. (2011). Adsorption removal of ammonium and phosphate from water by fertilizer controlled release agent prepared from wheat straw. *Chemical Engineering Journal*, 171(3), 1209–1217. <https://doi.org/10.1016/j.cej.2011.05.027>
- Marañón, E., Ulmanu, M., Fernández, Y., Anger, I., & Castrillón, L. (2006). Removal of ammonium from aqueous solutions with volcanic tuff. *Journal of Hazardous Materials*, 137(3), 1402–1409. <https://doi.org/10.1016/j.jhazmat.2006.03.069>
- Metcalf & Eddy. (2003). Wastewater Engineering: Treatment and Reuse (Book). In *Chemical engineering* (Nomor 7, hal. 10–11).

- Minamisawa, M., Minamisawa, H., Yoshida, S., & Takai, N. (2004). Adsorption behavior of heavy metals on biomaterials. *Journal of Agricultural and Food Chemistry*, 52(Ii), 5606–5611.
- Mustofa, A. (2015). Kandungan nitrat dan pospat sebagai faktor tingkat kesuburan perairan pantai. *Disprotek*, 6(1), 13–19.
- Omidire, N., Shange, R., Khan, V., Bean, R., & Bean, J. (2015). Assessing the Impacts of Inorganic and Organic Fertilizer on Crop Performance Under a Microirrigation-Plastic Mulch Regime. *Professional Agricultural Workers Journal*, 3(1), 6. <https://doi.org/10.3390/su13063150>
- Ozturk, N. (2004). Nitrate removal from aqueous solution by adsorption onto various materials. 112, 155–162. <https://doi.org/10.1016/j.jhazmat.2004.05.001>
- Pane, F. A. (2019). Studi pengolahan air limbah tahu secara anaerob dengan media bioball dan fitoremediasi oleh tanaman kiambang (*salvinia molesta*) dalam penurunan COD dan TSS. 12–61.
- Pemerintah, P. (2021). Peraturan Pemerintah Republik Indonesia Nomor 22 Tahun 2021 Tentang Penyelenggaraan Perlindungan dan Pengelolaan Lingkungan Hidup. *Sekretariat Negara Republik Indonesia*, 1(078487A), 483. <http://www.jdih.setjen.kemendagri.go.id/>
- Purmaningtyas, S. E. (2014). Distribusi Konsentrasi Oksigen, Nitrogen dan Fosfat di Waduk Saguling, Jawa Barat. *Limnotek*, 21(2), 125–134.
- Putri, V. I. (2017). Pemberian Beberapa Jenis Biochar Untuk Memperbaiki Sifat Kimia Tanah Ultisol Dan Pertumbuhan Tanaman Jagung. *Agroekoteknologi*, 5(4), 824–828. <https://doi.org/10.32734/jaet.v5i4.16435>
- Salim, H. (2002). Beban Pencemaran Limbah Domestik dan Pertanian di DAS Citarum Hulu. *Jurnal Teknologi Lingkungan*, 3(2), 107–111. <http://www.kelair.bppt.go.id/Jtl/2002/vol3-2/04cemar.pdf>
- Saragih, S. . (2008). *Pembuatan dan Karakterisasi Karbon Aktif dari Batubara Riau sebagai Adsorben*. Universitas Indonesia.
- Setiowati, Roto, & Wahyuni, E. T. (2016). Monitoring Kadar Nitrit Dan Nitrat Pada Air Sumur Di Daerah Catur Tunggal Yogyakarta Dengan Metode Spektrofotometri Uv-Vis. *Jurnal Manusia dan Lingkungan*, 23(2), 143–148.
- Singh, N. B., Nagpal, G., Agrawal, S., & Rachna. (2018). Water purification by using Adsorbents: A Review. *Environmental Technology and Innovation*, 11, 187–240. <https://doi.org/10.1016/j.eti.2018.05.006>
- Timnie R, P. (2018). *Pemanfaatan kulit durian sebagai adsorben untuk penyisihan detergen dan fosfat dalam pengolahan limbah cair laundry*.

- Triyono, A., Purwanto, & Budiyo. (2013). Efisiensi Penggunaan Pupuk – N untuk Pengurangan Kehilangan Nitrat pada Lahan Pertanian. *Prosiding Seminar Nasional Pengelolaan Sumber Daya Alam dan Lingkungan*, 1, 526–531.
- Uchimiya, M., Chang, S. C., & Klasson, K. T. (2011). Screening biochars for heavy metal retention in soil: Role of oxygen functional groups. *Journal of Hazardous Materials*, 190(1–3), 432–441. <https://doi.org/10.1016/j.jhazmat.2011.03.063>
- Wan, S., Wang, S., Li, Y., & Gao, B. (2016). Functionalizing biochar with Mg–Al and Mg–Fe layered double hydroxides for removal of phosphate from aqueous solutions. *Journal of Industrial and Engineering Chemistry*, 47, 246–253. <https://doi.org/10.1016/j.jiec.2016.11.039>
- Wang, B., Lehmann, J., Hanley, K., Hestrin, R., & Enders, A. (2015). Adsorption and desorption of ammonium by maple wood biochar as a function of oxidation and pH. *Chemosphere*, 138, 120–126. <https://doi.org/10.1016/j.chemosphere.2015.05.062>
- Wang, Z., Guo, H., Shen, F., Yang, G., Zhang, Y., Zeng, Y., Wang, L., Xiao, H., & Deng, S. (2015). Chemosphere Biochar produced from oak sawdust by Lanthanum (La)-involved pyrolysis for adsorption of ammonium (NH₄⁺), nitrate (NO₃⁻), and. *Chemosphere*, 119, 646–653. <https://doi.org/10.1016/j.chemosphere.2014.07.084>
- Warlina, L. (2004). *Pencemaran Air* : 1–26.
- Wei, D., Li, B., Huang, H., Luo, L., Zhang, J., Yang, Y., Guo, J., Tang, L., Zeng, G., & Zhou, Y. (2018). Biochar-based functional materials in the purification of agricultural wastewater: Fabrication, application and future research needs. *Chemosphere*, 197, 165–180. <https://doi.org/10.1016/j.chemosphere.2017.12.193>
- Widwastuti, H., Bisri, C., & Rumhayati, B. (2019). *Pengaruh Massa Adsorben dan Waktu Kontak terhadap Adsorpsi Fosfat menggunakan Kitin Hasil Isolasi dari Cangkang Udang*. 93–98.
- Xiang, W., Zhang, X., Chen, J., Zou, W., He, F., Hu, X., Tsang, D. C. W., Ok, Y. S., & Gao, B. (2020). Biochar technology in wastewater treatment: A critical review. *Chemosphere*, 252, 126539. <https://doi.org/10.1016/j.chemosphere.2020.126539>
- Xu, K., Lin, F., Dou, X., Zheng, M., Tan, W., & Wang, C. (2018). Wood waste Mixing. *Journal of Cleaner Production*. <https://doi.org/10.1016/j.jclepro.2018.03.206>
- Xue, L., Gao, B., Wan, Y., Fang, J., Wang, S., Li, Y., Muñoz-carpena, R., & Yang, L. (2016). High efficiency and selectivity of MgFe-LDH modified wheat-straw biochar in the removal of nitrate from aqueous solutions. *Journal of the*

Taiwan Institute of Chemical Engineers, 0, 1–6.
<https://doi.org/10.1016/j.jtice.2016.03.021>

- Yakout, S. M., Daifullah, A. E. H. M., & El-Reefy, S. A. (2015). Pore Structure Characterization of Chemically Modified Biochar Derived From Rice Straw. *Environmental Engineering and Management Journal*, 14(2), 473–480. <https://doi.org/10.30638/eemj.2015.049>
- Yang, H. I., Lou, K., Rajapaksha, A. U., Ok, Y. S., Anyia, A. O., & Chang, S. X. (2017). Adsorption of ammonium in aqueous solutions by pine sawdust and wheat straw biochars. *Environmental Science and Pollution Research*, 25(26), 25638–25647. <https://doi.org/10.1007/s11356-017-8551-2>
- Yang, L., Yang, M., Xu, P., Zhao, X., Bai, H., & Li, H. (2017). Characteristics of nitrate removal from aqueous solution by modified steel slag. *Water (Switzerland)*, 9(10). <https://doi.org/10.3390/w9100757>
- Yin, Q., Ren, H., Wang, R., & Zhao, Z. (2018). Evaluation of nitrate and phosphate adsorption on Al-modified biochar: Influence of Al content. *Science of the Total Environment*, 631–632, 895–903. <https://doi.org/10.1016/j.scitotenv.2018.03.091>
- Zhang, M., Gao, B., Yao, Y., Xue, Y., & Inyang, M. (2012). Synthesis of porous MgO-biochar nanocomposites for removal of phosphate and nitrate from aqueous solutions. *Chemical Engineering Journal*, 210, 26–32. <https://doi.org/10.1016/j.cej.2012.08.052>
- Zhang, Y. L. Z. I. (2014). *Recovery of NH₄⁺ by corn cob produced biochars and its potential application as soil conditioner.*
- Zhao, H., Xue, Y., Long, L., & Hu, X. (2018). Adsorption of nitrate onto biochar derived from agricultural residuals. *Water Science and Technology*, 77(2), 548–554. <https://doi.org/10.2166/wst.2017.568>
- Zhou, L., Xu, D., Li, Y., Pan, Q., Wang, J., Xue, L., & Howard, A. (2019). Phosphorus and nitrogen adsorption capacities of biochars derived from feedstocks at different pyrolysis temperatures. *Water (Switzerland)*, 11(8), 1–16. <https://doi.org/10.3390/w11081559>
- Zhou, Y., Zhang, F., Tang, L., Zhang, J., Zeng, G., Luo, L., Liu, Y., Wang, P., Peng, B., & Liu, X. (2017). Simultaneous removal of atrazine and copper using polyacrylic acid-functionalized magnetic ordered mesoporous carbon from water: Adsorption mechanism. *Scientific Reports*, 7(March), 1–10. <https://doi.org/10.1038/srep43831>
- Zulkarnaini. (2020). *Penemuan dan Aplikasi Anammox.* Padang.