PENGARUH ASAM LEMAH SEBAGAI PELARUT CaO DALAM SINTESIS HIDROKSIAPATIT DARI CANGKANG KERANG DARAH (Anadara granosa) SEBAGAI PENYERAP ZAT WARNA MALACHITE GREEN



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ABSTRACT

THE EFFECT OF WEAK ACIDS AS A SOLVENT OF CaO IN THE SYNTHESIS OF HYDROXYAPATITE FROM BLOOD CRUSH SHELLS (*Anadara granosa*) AS AN ABSORBENT OF MALACHITE GREEN DYE

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The dissolution of CaO from blood cockle shells using weak acid solvents has been successfully carried out to synthesize hydroxyapatite (HAp) through the sol-gel method. The variations of weak acids used include acetic acid, ascorbic acid, citric acid, and oxalic acid. Among the four variations, HAp synthesized using citric acid (HAp-citrate) showed the best performance with the highest yield of 28.44% and ad<mark>so</mark>rp<mark>tion efficienc</mark>y against Malachite Green (MG) dye reaching 90.5<mark>6%. Fourier</mark> Transform Infrared (FTIR) characterization showed that all samples had PO₄³⁻ and OH⁻ functional groups which are typical groups of HAp. The results of X-Ray Diffraction (XRD) analysis showed the crystallite size of each HAp was 11.91 nm (HAp-acetate), 12.97 nm (HAp-ascorbate), 13.52 nm (HAp-citrate), and 15.35 nm (HAp-oxalate). The morphology of the Scanning Electron Microscopy-Energy Dispersive Spectroscopy (SEM-EDS) results showed irregular particle shapes and agglomerati<mark>on, with</mark> the average particle sizes of HAp-acetate, HAp-ascorbate, HAp-citrate and HAp-oxalate respectively being 1.58; 2.22; 2.07; and 2.96 µm and the Ca/P ratio of each sample was obtained at 1.16; 1.29; 1.38; and 1.31. Surface Area Analyzer (SAA) analysis showed differences in surface area between samples. Adsorption isotherm studies showed that HAp-citrate followed the Freundlich model and pseudo-second-order kinetics. In addition, reusability tests showed that HAp-citrate could be reused up to five cycles without signific<mark>an</mark>t decrease in adsorption efficiency. Overall, HAp synthesized using weak acid solvents has high po<mark>te</mark>ntial as an ad<mark>sorb</mark>ent mate<mark>rial in eff</mark>orts to reduce MG dye waste from the envir<mark>on</mark>ment.

Keywords: Hydroxyapatite, Weak acids, Blood clam shell, Adsorption, Malachite green